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A chiral (alkoxy)methyl-substituted silicon group as an auxiliary for the stereoselective cuprate addition to α,β -unsaturated ketones: synthesis of (-)-(R)-phenyl 2-phenylpropyl ketone

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Abstract: The [(benzyloxy)methyl](tert-butyl)methylsilyl group was used as the chiral auxiliary to effect highly diastereoselective conjugate additions of organocuprates to enones. Thus, reaction of several R₂CuLi with α -silylated α , β -unsaturated ketones afforded the respective addition products with π -face selectivities of up to 99%. Starting with an optically active substrate, enantiomerically enriched (–)-(R)-phenyl 2-phenylpropyl ketone was prepared with virtually no loss of chiral information in a reaction sequence involving cuprate addition, hydrolysis, and removal of the silicon group. © 1997 Elsevier Science Ltd

Introduction

The conjugate addition of organocuprates and other organometallics to α,β -unsaturated ketones and carboxylic acid derivatives is widely used in organic synthesis for the preparation of specifically substituted carbonyl and carboxyl compounds. In connection with the increasing importance of synthetic access to products of stereochemical integrity, the quest for stereoselective variations of this reaction started already some years ago. So far, a number of stereoselective transformations have been elaborated, including enantioselective processes that make use of chiral auxiliaries such as, e.g., chiral sulfoxides or chiral alkoxy or alkylamino/amido groups. Whereas 1,4-additions to chiral ester and amide derivatives of α,β -unsaturated carboxylic acids proceed often with high degrees of diastereoselectivity, the stereochemical success of reactions with α -auxiliary-substituted α,β -unsaturated ketones, particularly in the cases of acyclic enone systems, has yet remained rather modest.

In this paper we present the use of the chiral silicon group A as the stereochemical director for reactions of organocuprates with α,β -unsaturated ketones. The group A has already provided high diastereoselectivities in chelate controlled 1,2-additions to acylsilanes^{11,12} and, as shown below, proved equally potent as chiral auxiliary for stereoselective 1,4-additions.

Results and discussion

The model compounds of the type 5a,b and the chiral α -silyl-substituted α,β -unsaturated ketones of the type 6a,b used for this study have been prepared in a straightforward manner from the corresponding chlorosilanes 1 and 2 via the allylic alcohols of the type 3a,b and 4a,b (Scheme 1). The syntheses of the latter compounds has been described before. Oxidation of the alcohols (E)-3a,b and (E)-4a,b

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with *Jones* reagent ¹⁴ and of the alcohols (Z)-3a,b and (Z)-4a¹⁵ with MnO₂ afforded the respective isomerically pure α, β -unsaturated ketones 5a,b and 6a,b in high yields. The use of milder oxidation conditions for the preparation of the (Z)-configured compounds 5a,b and 6a was necessary because the reaction of the respective precursors, *e.g.*, (Z)-4a, with CrO₃ provided substantial amounts of over-oxidized α, β -epoxyketones, *e.g.*, 7a. The reason for this peculiar behavior of the silyl-substituted (Z)-configured allyl alcohols is not known.

The structures of the compounds **5** and **6**, particularly their double bond geometries, were ascertained by correlation with precursor molecules. Additionally, the vinylic protons show absorptions in the 1 H-NMR spectra characteristically different for the (E)- and (Z)-configured enones, the chemical shifts being moved towards lower fields for the latter compounds. Sound evidence for the assigned double bond geometries arose also from the values of $^{3}J(^{29}\text{Si},^{1}\text{H})$ -coupling constants determined for the vicinal couplings of the vinylic Si- and H nuclei with the novel ACT-J-NMR experiment: 16 the (Z)-configured compounds displayed $^{3}J(^{29}\text{Si},^{1}\text{H})_{trans}$ -values of 13–15 Hz, the (E)-configured analogs $^{3}J(^{29}\text{Si},^{1}\text{H})_{cis}$ -values of 7–8 Hz. These values are characteristic for $^{3}J(^{29}\text{Si},^{1}\text{H})$ -coupling constants of vinylsilanes of the given substitution patterns. 17

Cuprate additions to α -silylated α,β -unsaturated ketones were initially tested and optimized with the achiral model compounds **5a,b**. Best results were obtained when the enones of the type **5** were treated at low temperature with the respective lithium diorganylcuprates, and when the resulting enolates were trapped as the silyl enol ethers (*E*)- or (*Z*)-**8** (Scheme 2, Table 1). Hydrolysis with aqueous H₂SO₄ in acetone provided the epimeric α -silylated ketones **10/10'**, which gave the desilylated ketone **12a** (R², R³=Me, Ph) upon treatment with TBAF in MeCN.

It was imperative to capture the initially formed cuprate addition products as enol ethers: direct hydrolytic workup led to the formation of substantial amounts of oxidized silicon-free side products. The addition of N, N, N', N'-tetramethylethylendiamine (TMEDA) to the enolate/TMSCl solution was neccessary to ensure complete silylation.

The reactions of the enones (E)-5b and (Z)-5a,b with approximately 1.5 equivalents of organocuprates and TMSCl to afford the silyl enol ethers of the type 8 were highly stereoselective: starting from the (E)-configured substrate (E)-5b, the respective (E)-configured enol ether 8 was obtained as the major product, and departing from the (Z)-configured starting materials (Z)-5a,b, the (Z)-configured products of the type 8 were formed with high preference (E) (Table 1). The selectivities probably reflect the preferred conformations of the starting materials, being most probably s-cis for (E)-5a,b and s-trans

1)
$$R_{2}^{3}$$
 CULI
 $Et_{2}O / -80 \, ^{\circ}C$

2) TMSCI/TMEDA
 $-80 \, ^{\circ}C$

(E)- or (Z)-5a,b
(E)- or (Z)-6a,b
(E)- and (Z)-8
(E)- and (Z)-9a-e

TBAF / MeCN

TMSO
 R_{1}^{1} Ph
 R_{3}^{2}

TMSCI/TMEDA
 R_{3}^{2}

(E)- and (Z)-8
(E)- and (Z)-9a-e

TBAF / MeCN

For R¹, R², R³, double bond configuration, and diastereoselectivities see the *Table*.

Scheme 2.

for (Z)-5a,b; ¹⁹ they are not considered to be the direct consequence of the double bond geometries of the enones. This assumption is important in connection with the interpretation of the stereoselectivities found in the cuprate additions performed with the chiral silicon compounds of the type 6. Among the chiral starting materials 6a,b, the ketones (E)-6a and (Z)-6a delivered the respective addition products (SiR^*,S^*,Z) -9a-c and (SiR^*,R^*,Z) -9a-c with high stereoselectivities, not only in respect to the newly formed stereogenic center at C(3) but also concerning the double bond configuration of the enol ether. The selectivities anent chiral induction of the reactions with the (Z)-configured compound were virtually complete whereas those of the transformations of the (E)-configured analog were slightly lower, though still satisfactory. The selectivities attained in the preparation of compounds of the type 9 starting from the phenyl-substituted (E)-6b, however, were disappointingly poor in all respects.

The high π -face selectivities of the cuprate additions to the enones (E)- and (Z)-6a were attributed to intermediary chelate-structures of the type 13a (Scheme 3). It is assumed that such species are attacked by the copper reagents from the least hindered site, opposite to the tert-butyl group. This gives rise to (Z)-enolates of the type 14, which are subsequently trapped by the silvlating agent. In the case of (Z)-6a, the respective stereochemically crucial intermediate (Z)-13a should be formed particularly facile since negligible steric constraints are introduced into the structure. This is not the case with (E)-6a, where A_{1,3} strain arises when the enone is forced from the preferred s-cis into the s-trans conformation. Therefore, it might be assumed that (E)-6a will not exclusively react via the highly organized intermediary (E)-13a, and this would thus account for the lower selectivities found with (E)-6a as the starting material. The still prominent diastereoselectivities obtained in the cuprate additions to (E)-6a show, however, that the $A_{1,3}$ strain in complex (E)-13a is not too high and can be mostly overcome by the returns of free energy attained by the formation of the chelate. With increasing size of a vinylic substituent positioned cis to the benzoyl group in compounds of the type (E)-6, though, increasing $A_{1,3}$ strain is introduced into a derived chelate structure of the type (E)-13. As a result, chelates of the type (E)-13 should become less favorable and the stereoselectivities of conjugate additions must be expected to decrease. The poor stereochemical results of the reactions of organocuprates with (E)-6b, having the rather large phenyl group positioned cis to the benzoyl moiety, are in fact attributed to this effect.

That chelate intermediates of the type 13 might be important for the stereochemical course of the

Table 1. Organocuprate additions to enones of the type 5 and 6, hydrolysis, and desilylation

Starting	Starting Materials			Products								
Fnone			Cuprate	Silyl Enol Ether			Silylated Ketone			İ	Final Ketone	etone
	-	20	D3	No.		yield	(qoN	ratio	્ જુ ફુંટ જુ	yield (%) ^d)	No	yield (%)
ON !	·	2	2 2	0 (2)/0	(2.1)8)	76	10/10,	-		99	12a	82
(E)-5a	Me	Me	Ξ		(/ 1			-			ı	١
(Z)-5a	Me	Σe	Ph	(E)-8/(Z)-8 (1)	(1:10)	91	10/10′				İ	
(E)- Sb	Me	몺	Me	(E)-8/(Z)-8 (>	(>40:1)	93	-	1	1	l		1
16	Me	4	Me	(E)-8/(Z)-8 (1)	(1:6)	2		-	ļ		1	1
(E)-6a	BnOCH ₂		Ph	<u>~</u>	(SiR*,S*,Z)-9a	62	(SiR*,3R*)-11a/(SiR*,3S*)-11a (SiR*,3R*)-11a'/(SiR*,3S*)-11a'	0.5 : 93 0.5 : 6	86	76	12a	87
			ŭ	(SiR*,R*,Z)-9b/(SiR*,S*,Z)-9b	(SiR*,S*,Z)-9b	73	(SiR*,3R*)-11b/(SiR*,3S*)-11b (SiR*,3R*)-11b'/(SiR*,3S*)-11b'	5 : 31 / 5 : 59	80	82	12b	98
			Bu	(SiR*,R*,Z)-9c/(SiR*,S*,Z)-9c	(SiR*,S*,Z)-9c	73	(SiR*,3R*)-11c/(SiR*,3S*)-11c (SiR*,3R*)-11c'/(SiR*,3S*)-11c'	4 : 28 / 5 : 63	82	91	12c	68
e9 -(Z)	(Z)-6a BnOCH ₂ Me	Me	F.	(SiR*,R*,Z)-9a/(SiR*,S*,Z)-9a	(SiR*,S*,Z)-9a	89	(SiR*,3R*)-11a/(SiR*,3S*)-11a (SiR*,3R*)-11a'/(SiR*,3S*)-11a'	20 : 2 78 : 0	96	94	I	1
			茁	(SiR*,R*,Z)-9b/(SiR*,S*,Z)-9b	'(SiR*,S*,Z)-9 b	78	(SiR*,3R*)-11b/(SiR*,3S*)-11b (SiR*,3R*)-11b'/(SiR*,3S*)-11b'	38 : 3 / 58 : 1	92	68	1	1
			Bu	(SiR*,R*,Z)-9c/(SiR*,S*,Z)-9c	(SiR*,S*,Z)-9c	93	(SiR*,3R*)-11c/(SiR*,3S*)-11c (SiR*,3R*)-11c'/(SiR*,3S*)-11c'	46 : 2.5 <i>/</i> 48 : 3.5	88	68	1	
(E)-6b	(E)-6b BnOCH ₂ Ph	P.	Me	(SiR*,R*,E)-9a/(SiR*,R*,Z)-9a (SiR*,S*,E)-9a/(SiR*,S*,Z)-9a	(SiR*,R*,E)-9a/(SiR*,R*,Z)-9a (SiR*,S*,E)-9a/(SiR*,S*,Z)-9a	75	(SiR*,3R*)-11a/(SiR*,3S*)-11a (SiR*,3R*)-11a'/(SiR*,3S*)-11a'	19: 19 31: 31	0	68	1	l
			超	(SiR*,R*,E)-9d. (SiR*,S*,E)-9dv	(SiR*,R*,E)-9d/(SiR*,R*,Z)-9d (SiR*,S*,E)-9d/(SiR*,S*,Z)-9d	68	(SiR*,3R*)-11d/(SiR*,3S*)-11d (SiR*,3R*)-11d'/(SiR*,3S*)-11d'	28:4 51:17	28	75	12d	11
			Bu	(SiR*,R*,E)-9e. (SiR*,S*,E)-9e/	(SiR*,R*,E)-9e/(SiR*,R*,Z)-9e (SiR*,S*,E)-9e/(SiR*,S*,Z)-9e	59	(SiR*,3R*)-11e/(SiR*,3S*)-11e (SiR*,3R*)-11e ⁷ /(SiR*,3S*)-11e ³	7: 37	17	93	12e	73

This ratio cannot be compared directly to the ratios obtained with the other compounds of the type 5. The reaction had to be performed in presence of a large excess of Ph_2CuL_1 (6 eq.) to ensure complete conversion of (E)-5a. This was not necessary for the other compounds that reacted to full conversion already with approx. 1.5 eq. of the organocuprate. a (

We were not able to determine the relative configurations of the stereogenic centers at C(2). Groups of compounds with or without the suffix '9' possess the same relative configuration at the stereogenic centers on the carbon framework (for the compounds of the type 11b,c/11b,c' determined from similarity examples configuration at the stereogenic centers on the carbon framework (for the compounds of the type 11b,c'11b,c' determined from similarity configuration at the stereogenic centers on the carbon framework (for the compounds of the type 11b,c'11b,c' determined from similarity configuration at the stereogenic centers on the carbon framework (for the compounds of the type 11b,c'11b,c' determined from similarity configuration at the stereogenic centers on the carbon framework (for the compounds of the type 11b,c'11b,c' determined from similarity configuration at the stereogenic centers on the carbon framework (for the compounds of the type 11b,c'11b,c' determined from similarity configuration at the stereogenic centers on the carbon framework (for the compounds of the type 11b,c' determined from the carbon framework (for the compounds of the type 11b,c' determined from the carbon framework (for the compounds of the type 11b,c') and the configuration of the carbon framework (for the compounds of the type 11b,c') and the carbon framework (for the compounds of the type 11b,c') and the carbon framework (for the compounds of the type 11b,c') and the carbon framework (for the compounds of the type 11b,c') and the carbon framework (for the compounds of the type 11b,c') and the carbon framework (for the compounds of the type 11b,c') and the carbon framework (for the compounds of the type 11b,c') and the carbon framework (for the compounds of the type 11b,c') and the carbon framework (for the compounds of the type 11b,c') and the carbon framework (for the compounds of the type 11b,c') and the compound of the carbon framework (for the compounds of the compound of the compound of the compound of the compound of the comp ties in the 1H-NMR spectra). <u>ء</u>

Diastereomeric excess in respect to the stereogenic centers at silicon and C(3).

c) Diastereomeric excess in respect to the stereogenic centers at structure and except in situ hydrolyzed upon their formation.

d) Higher overall yields were obtained when the enol ethers of the type 8 or 9 were in situ hydrolyzed upon their formation.

BnO Ph

$$t = Bu - Si$$

Me

 $t = Bu - Si$

Me

 $t =$

Scheme 3.

cuprate additions to (E)-6a and (Z)-6a is supported by the fact that exclusively the (Z)-configured enol ethers 9a-c were formed. Evidently, and contradictory to the reactions with the achiral model compounds (E)-5a,b and (Z)-5a,b, a s-trans intermediate is involved not only in the transformation of the (Z)- but also of the (E)-configured 6a. This seems not to be the case in the transformation of (E)-6b: since mixtures of four compounds were obtained after its treatment with cuprates and silylation, it must be assumed that the increased steric strain resulting from a conformational change from the s-cis to the s-trans form of (E)-6b could only partially be overcome. Reduced stereoselectivities were notably also obtained when the cuprate additions to any of the compounds of the type 6 were performed in the presence of a donating additive (like, e.g., TMEDA or THF) that can compete as a ligand for complexation of the cations with the substrate and thus will disfavor the formation of the chelates of the type 13.

The stereochemistry of the cuprate additions follows the course that is anticipated on the basis of the above mentioned 'chelate model'. This is secured by the result obtained from a reaction sequence starting with optically active substrate (+)-(R,E)-6a. Enantiomerically enriched (+)-(R,E)-6a $(96\pm2\%$ ee) was prepared from acetylsilane (-)-(R)-15 $(96\pm2\%$ ee)^{20,21} by a procedure described earlier. The compound was treated with Ph₂CuLi and led, after direct hydrolysis of the intermediary silyl enol ether (SiR,3S,Z)-9a, in high yields to the respective mixtures of epimeric ketones (SiR,3S)-11a/11a' (Scheme 4). Desilylation, which was performed likewise with the racemic ketones of the type 11 and 11' (giving racemic 12a-e, Scheme 2, Table 1), was effected by treatment of (SiR,3S)-11a/11a' with TBAF in MeCN. The compound obtained this way in 83% overall yield (from (+)-(R,E)-6b) and in 94±2% ee was found to be (-)-(R)-phenyl 2-phenyl propyl ketone ((-)-(R)-12a), as assigned on the basis of its spectroscopic and chiroptic properties. The configuration of the chiral center of (-)-(R)-12a, thus, corresponds to an attack of the cuprate to the (-)-side of the (-)-system.

Scheme 4.

The formation of (-)-(R)-12a from (+)-(R,E)-6a does at this instance not only support the above proposed chelate model for the stereoselective conjugate cuprate addition to chiral α -silyl-substituted α,β -unsaturated ketones, but the synthesis of an acyclic β -chiral ketone in an enantiomeric excess of 94±2% by means of an auxiliary-assisted diastereoselective cuprate addition represents also one of the best examples of such a process. Thus, we have shown that the chiral silicon group **A** can efficiently be used as a stereochemical director in 'chelate-controlled' conjugate addition reactions.

Experimental

General remarks

Unless otherwise stated: all org. solvents were distilled prior to use. For the reactions, THF and Et₂O were dried over Na in presence of diphenylketyl; CH₂Cl₂ was dried over molecular sieves (3 Å). All reactions were carried out under an Ar atmosphere. Soln. of salts and acids for workup procedures were prepared in deionized H₂O. Extracts were dried (MgSO₄) and evaporated *in vacuo*. Chromatography: silica gel (SiO₂) Merck 60 (40–63 µm). M.p.: Mettler FP-5/FP-52. IR (neat): Perkin-Elmer 781; data in cm⁻¹. UV/VIS (CHCl₃): Perkin Elmer 555; λ_{max} in nm (lg ε). ¹H-NMR: at 300 MHz in CDCl₃; Bruker AC-300 or Bruker ARX-300; δ in ppm rel. to CHCl₃ (=7.26 ppm); *J* in Hz. ¹³C-NMR: at 75.6 MHz in CDCl₃; Bruker ARX-300; δ in ppm rel. to CDCl₃ (=77.0 ppm); multiplicities from DEPT experiments. ²⁹Si-NMR: at 119.2 MHz in CDCl₃; Bruker ARX-600; δ in ppm rel. to SiMe₄ (=0.00 ppm). ¹⁹F-NMR: at 564.5 MHz in CDCl₃; Bruker ARX-600; δ in ppm rel. to CCl₃F (=0.00 ppm). CI–MS (chemical ionization mass spectrometry) with NH₃ as the reactant gas; Finnigan SSQ 700 or Varian MAT 90; base peak and quasi-molecular ions only; data in m/z.

1. Oxidation of the allylic alcohols of the type 3 and 4

1.1. General procedures

A: To a soln. of the respective allylic alcohol in acetone (0.1 M) was added dropwise Jones reagent ¹⁴ until the color of the mixture remained persistently brown. It was diluted with H₂O, neutralized with sat. aq. NaHCO₃ soln., extracted with Et₂O, and chromatographed (SiO₂, hexane/EtOAc 25:1). B: To a soln. of the respective allylic alcohol in CH₂Cl₂ (0.1 M) was added MnO₂ (5 eq.), and the mixture was refluxed for 2 h. It was filtered, the solvent was evaporated, and the crude product chromatographed (SiO₂, hexane/EtOAc 25:1).

1.2. (E)-1-[(tert-Butyl)dimethylsilyl]prop-1-enyl phenyl ketone (E)-5a

According to 1.1, procedure A: (E)-**3a**¹³ (232 mg, 0.89 mmol) was converted to (E)-**5a** (201 mg, 0.77 mmol, 87%). IR: 3080w, 3060w, 3020w, 2959s, 2920s, 2880s, 2850s, 1655s, 1595s, 1575s, 1470m, 1460m, 1445s, 1410m, 1390m, 1370m, 1360m, 1345s, 1310w, 1245s, 1225s, 1170s, 1125w, 1070w, 1030m, 1015s, 1005m, 970w, 935w, 870m, 835s, 820s, 805s, 785m, 765s, 740m, 710s, 690s. UV/VIS: 242 (3.98). ¹H-NMR: 7.94–7.45 (m, 5 arom. H); 6.25 (q, J=6.7, HC=); 1.63 (d, J=6.7, MeC=); 0.91 (s, t-Bu); 0.05 (s, Me₂Si). ¹³C-NMR: 202.3 (s, C=O); 145.3 (s, SiC=); 140.1 (d, HC=); 137.6 (s, arom. C); 133.1 (d, arom. C); 129.4, 128.8 (2d, 2×2 arom. C); 27.7 (q, Me_3 C); 18.3 (q, MeCH); 18.0 (s, Me₃C); -5.5 (q, MeSi). CI–MS: 261 [M+H]⁺.

1.3. (Z)-1-[(tert-Butyl)dimethylsilyl]prop-1-enyl phenyl ketone (Z)-5a

According to 1.1, procedure B: (Z)- $3a^{13}$ (394 mg, 1.50 mmol) was converted to (Z)-5a (370 mg, 1.42 mmol, 95%). IR: 3070w, 3050w, 3020w, 2950s, 2920s, 2890s, 2850s, 1650s, 1595s, 1575m, 1465m, 1460m, 1445s, 1405m, 1390m, 1370m, 1340w, 1310m, 1250s, 1175m, 1155w, 1130w, 1070w, 1050m, 1035w, 1020m, 1005m, 955w, 935w, 905w, 870w, 835s, 825s, 800s, 760s, 740w, 730w, 705s. UV/VIS: 245 (4.02). ¹H-NMR: 7.82–7.39 (m, 5 arom. H); 6.57 (q, J=6.6, HC=); 1.99 (d, J=6.6, MeC=); 1.01 (s, t-Bu); 0.17 (s, Me₂Si). ¹³C-NMR: 201.5 (s, C=O); 147.7 (d, HC=); 143.1 (s, SiC=); 138.2 (s, arom. C); 132.2 (d, arom. C); 129.9, 128.1 (2d, 2×2 arom. C); 27.2 (q, Me_3 C); 18.8 (q, MeCH); 18.5 (s, Me₃C); -3.6 (q, MeSi). CI–MS: 261 [M+H]⁺.

1.4. (E)-1-[(tert-Butyl)dimethylsilyl]-2-phenylethenyl phenyl ketone (E)-5b

According to 1.1, procedure A: (E)-3b¹³ (113 mg, 0.35 mmol) was converted to (E)-5b (98 mg, 0.30 mmol, 87%). IR: 3080w, 3050m, 3020m, 2950s, 2930s, 2890m, 2880m, 2850s, 1700m, 1650s, 1595m, 1580m, 1490w, 1465m, 1469m, 1445m, 1410w, 1390m, 1360m, 1315m, 1280w, 1265m, 1250m, 1230s, 1200m, 1175m, 1070w, 1045m, 1005m, 935m, 925m, 900m, 880m, 835s, 820s, 710s. UV/VIS: 247 (4.30). ¹H-NMR: 7.81-7.04 (m, 10 arom. H); 6.98 (s, HC=); 0.91 (s, t-Bu); 0.07 (s, Me₂Si). ¹³C-NMR: 202.1 (s, C=O); 145.2 (s, arom. C); 141.3 (d, HC=); 138.6 (s, SiC=); 136.3 (s, arom. C); 132.8 (d, arom. C); 129.1, 128.7 (2d, 2×2 arom. C); 128.3, 128.2 (2d, 2×2 arom. C); 128.1 (d, arom. C); 26.7 (g, Me₃C); 18.1 (s, Me₃C); -5.7 (g, Me₂Si). CI-MS: 323 [M+H]⁺.

1.5. (Z)-1-[(tert-Butyl)dimethylsilyl]-2-phenylethenyl phenyl ketone (Z)-5b

According to 1.1, procedure B: (*Z*)-**3b**¹³ (220 mg, 0.68 mmol) was converted to (*Z*)-**5b** (187 mg, 0.58 mmol, 86%). IR: 3080w, 3060m, 3020w, 2950s, 2930s, 2890m, 2850s, 1655s, 1595m, 1580m, 1485m, 1470m, 1460m, 1445m, 1405w, 1390w, 1360w, 1310m, 1250s, 1235s, 1205w, 1175m, 1160w, 1115w, 1060m, 1025m, 1005w, 1000w, 935w, 910m, 870w, 835s, 820s, 790w, 760s, 750s, 735m, 700s. UV/VIS: 245 (4.20). ¹H-NMR: 7.78–7.11 (*m*, 10 arom. H); 7.33 (*s*, HC=); 0.69 (*s*, *t*-Bu); -0.27 (*s*, Me₂Si). ¹³C-NMR: 201.7 (*s*, C=O); 149.6 (*d*, HC=); 145.2 (*s*, arom. C); 138.1 (*s*, SiC=); 137.7 (*s*, arom. C); 132.6 (*d*, arom. C); 130.1, 128.3 (2*d*, 2×2 arom. C); 128.2 (*d*, arom. C); 127.9, 127.8 (2*d*, 2×2 arom. C); 27.6 (*q*, Me₃C); 18.1 (*s*, Me₃C); -2.8 (*q*, Me₂Si). CI–MS: 323 [*M*+H]⁺.

1.6. (\pm) -(E)- and (+)-(R,E)-1- $\{[(Benzyloxy)methyl](tert-butyl)methylsilyl\}prop-1$ -enyl phenyl ketone (E)-6a and (+)-(R,E)-6a

According to 1.1, procedure A: (*E*)-4a¹³ (440 mg, 1.20 mmol) or (*R,E*)-(-)-4a (320 mg, 0.87 mmol) was converted to (*E*)-6e (430 mg, 1.17 mmol, 98%) or (*R,E*)-(+)-6a (295 mg, 0.81 mmol, 93%). [α]_D²³=13.0 (c=1.7, THF), for (+)-(*R,E*)-6a. IR: 3080w, 3060w, 3020w, 2950s, 2920s, 2880m, 2850s, 1660s, 1590m, 1575m, 1490w, 1460m, 1445m, 1410w, 1390w, 1375m, 1360m, 1345m, 1310w, 1265m, 1250m, 1230s, 1170m, 1105m, 1095m, 1070m, 1030m, 1015m, 1000m, 980m, 935m, 900w, 865m, 825s, 810m, 780m, 765m, 735m, 720m, 695s. UV/VIS: 245 (4.03). ¹H-NMR: 7.85-7.12 (*m*, 10 arom. H); 6.25 (*q*, *J*=6.8, HC=); 4.22 (*s*, PhC H_2 O); 3.19 (*s*, SiCH₂O); 1.51 (*d*, *J*=6.8, MeC=); 0.86 (*s*, *t*-Bu); 0.01 (*s*, MeSi). ¹³C-NMR: 201.5 (*s*, C=O); 142.6 (*s*, arom. C); 141.1 (*d*, HC=); 138.7 (*s*, SiC=); 137.4 (*s*, arom. C); 132.8 (*d*, arom. C); 129.2, 128.4 (2*d*, 2×2 arom. C); 128.1, 127.4 (2*d*, 2×2 arom. C); 127.2 (*d*, arom. C); 77.0 (*t*, PhCH₂O); 60.5 (*t*, SiCH₂O); 27.0 (*q*, *Me*₃C); 18.1 (*q*, *Me*C=); 17.7 (*s*, Me₃C); -8.5 (*q*, MeSi). CI-MS: 367 [*M*+H]⁺.

1.7. (Z)-1-{(Benzyloxy)methyl}(tert-butyl)methylsilyl}prop-1-enyl phenyl ketone (Z)-6a

According to 1.1, procedure B: (*Z*)- $4a^{13}$ (300 mg, 0.82 mmol) was converted to (*Z*)-6a (291 mg, 0.80 mmol, 97%). IR: 3080w, 3060m, 3020m, 2950s, 2920s, 2890m, 2850s, 2810m, 1725w, 1700m, 1650s, 1595m, 1575m, 1490w, 1470m, 1460m, 1445m, 1390m, 1375m, 1360m, 1345w, 1310m, 1250s, 1200w, 1175m, 1155w, 1130w, 1105m, 1090m, 1070s, 1055m, 1035w, 1025m, 1005w, 975w, 935w, 905m, 879w, 830s, 825s, 805m, 770m, 760s, 735s, 705s. UV/VIS: 243 (4.09). ¹H-NMR: 7.69–7.02 (m, 10 arom. H); 6.53 (q, J=7.1, HC=); 4.20 (s, PhCH₂O); 3.31, 3.25 (*AB*, J_{AB} =12.6, SiCH₂O); 1.89 (*d*, J_{AB} =12.6, SiCH₂O); 1.89 (*d*, J_{AB} =12.6, SiCH₂O); 138.3 (s, arom. C); 131.9 (*d*, arom. C); 129.9, 128.0 (2*d*, 2×2 arom. C); 127.9, 127.4 (2*d*, 2×2 arom. C); 127.1 (*d*, arom. C); 77.1 (*t*, PhCH₂O); 61.3 (*t*, SiCH₂O); 27.7 (*q*, Me_{AB} C); 18.9 (*q*, Me_{AB} CH); 18.5 (*s*, Me_{AB} C); -6.2 (*q*, Me_{AB} C). CI-MS: 367 [*M*+H]⁺.

1.8. (E)-1-[[(Benzyloxy)methyl](tert-butyl)methylsilyl]-2-phenylethenyl phenyl ketone (E)-6b According to 1.1, procedure A: (E)-4b¹³ (400 mg, 0.93 mmol) was converted to (E)-6b (380 mg,

0.89 mmol, 95%). IR: 3080w, 3060m, 3020m, 2950s, 2930s, 2880m, 2850s, 2810m, 1700m, 1650s,

1595*m*, 1580*m*, 1570*m*, 1490*w*, 1470*m*, 1460*m*, 1445*m*, 1390*w*, 1380*m*, 1360*m*, 1310*w*, 1280*w*, 1250*m*, 1225*s*, 1200*m*, 1170*m*, 1155*w*, 1105*m*, 1090*m*, 1070*m*, 1050*m*, 1020*m*, 1005*w*, 980*w*, 940*w*, 930*w*, 900*w*, 880*w*, 825*m*, 780*m*, 765*m*, 745*m*, 735*m*, 695*s*. UV/VIS: 247 (4.25). ¹H-NMR: 7.86–7.05 (*m*, 15 arom. H); 7.05 (*s*, HC=); 4.32 (*s*, PhC*H*₂O); 3.38, 3.33 (*AB*, *J_{AB}* =13.1, SiCH₂O); 0.99 (*s*, *t*-Bu); 0.13 (*s*, MeSi). ¹³C-NMR: 201.9 (*s*, C=O); 142.8 (*s*, arom. C); 142.4 (*d*, HC=); 138.6 (*s*, SiC=); 136.6, 136.3 (2*s*, 2 arom. C); 132.7, 129.2, 128.9, 128.2, 128.1, 127.4, 127.2 (7*d*, 15 arom. C); 77.1 (*t*, Ph*C*H₂O); 60.4 (*t*, SiCH₂O); 27.1 (*q*, *Me*₃C); 18.2 (*s*, Me₃C); -8.5 (*q*, MeSi). CI–MS: 429 [*M*+H]⁺.

2. Preparation of the cuprates

2.1. Ph₂CuLi, Bu₂CuLi, and Me₂CuLi

The respective organolithium reagent (2 eq., commercial) was added to a suspension of CuI (20 mg/ml) in Et₂O at -50°C. The temperature was slowly raised to 0°C (30 min), and stirring was continued for an additional 2 h.

2.2. Et₂CuLi

t-BuLi (1 eq., 1.7 M in hexane) was added dropwise to a soln. of EtI in Et₂O (0.3–0.4 M) at -78° C. The temperature was slowly raised to 23°C and stirring was continued for 2 h. The resulting soln. was dropwise added at -50° C to a suspension of CuI (0.5 eq., 20 mg/ml) in Et₂O, the temperature was slowly raised to 0°C (30 min), and stirring was continued for an additional 2 h.

3. Cuprate additions to the silvlated enones of the type 5 and 6

3.1. General procedure

To a soln. of cuprate reagent (5–10 M in Et_2O) at $-80^{\circ}C$ were subsequently added TMSCl (2 eq.) and the respective ketone (0.2–0.5 eq.). The temperature was raised to 0°C over a period of 30 min., and the mixture was stirred for 1 h. TMEDA (2 ml) was added and stirring continued for an additional 30 min. It was quenched with sat. aq. NH₄Cl soln. at 0°C, extracted with Et_2O , and chromatographed (SiO₂, hexane/CH₂Cl₂ 5:1).

3.2. (E)- and (Z)-2- ${\text{(tert-Butyl)}}$ dimethylsilyl}-1- ${\text{(trimethylsiloxy)}}$ -1,3-diphenylbut-1-ene (E)-8 and (Z)-8

According to 3.1, (*E*)-5a (150 mg, 0.58 mmol) was reacted with Ph₂CuLi (4.18 mmol) and TMSCl (1 ml, 8.36 mmol) to give an inseparable mixture of (*E*)-8 and (*Z*)-8 (180 mg, 76%, (*E*)-8/(*Z*)-8=2:1 determined by 1 H-NMR). Likewise, (*Z*)-5a (90 mg, 0.35 mmol) gave with Ph₂CuLi (0.69 mmol) and TMSCl (0.1 ml, 0.79 mmol) (*Z*)-8 (129 mg, 0.31 mmol, 91%); (*E*)-5b (150 mg, 0.47 mmol) gave with Me₂CuLi (0.56 mmol) and TMSCl (0.14 ml, 0.94 mmol) (*E*)-8 (180 mg, 0.44 mmol, 93%); (*Z*)-5b (100 mg, 0.31 mmol) gave with Me₂CuLi (0.46 mmol) and TMSCl (0.1 ml, 0.80 mmol) a mixture of (*E*)-8 and (*Z*)-8 (82 mg, 0.20 mmol, 64%, (*E*)-8/(*Z*)-8=1:6); (*Z*)-5b (112 mg, 0.35 mmol) gave with Me₂CuLi (2.00 mmol) and TMSCl (0.55 ml, 4.38 mmol) a mixture of (*E*)-8 and (*Z*)-8 (110 mg, 0.27 mmol, 77%, (*E*)-8/(*Z*)-8=1:2).

Data of (E)-8 (colorless oil). IR: 3080w, 3050w, 3020w, 2950s, 2920s, 2870s, 2850s, 1610m, 1585s, 1485m, 1460m, 1440m, 1405w, 1385w, 1375w, 1365w, 1355w, 1325w, 1305w, 1260s, 1250s, 1190w, 1135s, 1110m, 1095m, 1070w, 1055w, 1030w, 1005w, 990w, 935w, 920w, 885m, 850s, 840s, 820s, 810s, 785m, 775s, 765s, 725m, 700s. 1 H-NMR: 7.48-7.12 (m, 10 arom. H); 3.64 (q, J=7.0, Ph(Me)CH); 1.69 (d, J=7.0, Ph(Me)CH); 0.92 (s, t-Bu); -0.37, -0.42 $(2s, Me_2Si)$; -0.44 (s, Me_3Si) . 1 H, 1 H-NOE: irrad. at -0.44, responsive signals at 7.48-7.47, 7.30-7.12, and 3.64. 13 C-NMR: 157.5 (s, PhC=); 146.6, 139.5 (2s, 2 arom. C); 129.8, 127.3, 126.9, 126.8, 126.4, 123.9 (6d, 10 arom. C); 15.7 (s, SiC=); 38.4 (d, Ph(Me)CH); 27.1 (q, Me_3C) ; 18.5 (q, Ph(Me)CH); 18.1 (s, Me_3C) ; 0.0 (q, Me_3Si) ; -4.5, -4.9 $(2q, Me_2Si)$. CI-MS: 411 $[M+H]^+$.

Data of (Z)-8 (colorless oil). 1 H-NMR: 7.61–7.12 (m, 10 arom. H); 3.53 (q, J=7.2, Ph(Me)CH); 1.33 (d, J=7.2, Ph(Me)CH); 0.95 (s, t-Bu); 0.16 (s, MeSi); -0.07 (s, Me₃Si); -0.44 (s, MeSi). 1 H-1

NOE: irrad. at -0.07, responsive signals at 0.95, 0.16, and -0.44. 13 C-NMR: 158.0 (s, PhC=); 145.5, 138.1 (2s, 2 arom. C); 127.4, 126.8, 126.5, 126.4, 125.9, 125.8 (6d, 10 arom. C); 117.3 (s, SiC=); 38.1 (d, Ph(Me)CH); 27.4 (q, Me₃C); 18.3 (q, Ph(Me)CH); 17.4 (s, Me₃C); 0.0 (q, Me₃Si); -3.1, -3.6 (2q, Me₂Si). CI-MS: 411 [M+H]⁺.

3.3. (SiR*,R*,E)-, (SiR*,S*,E)-, (SiR*,R*,Z)-, and (SiR*,S*,Z)-2-{[(benzyloxy)methyl](tert-butyl)-methylsilyl}-1-(trimethylsiloxy)-1,3-diphenylbut-1-ene (SiR*,R*,E)-9a, (SiR*,S*,E)-9a, (SiR*,S*,E)-9a, and (SiR*,S*,Z)-9a

According to 3.1, (E)-6a (100 mg, 0.27 mmol) was reacted with Ph₂CuLi (1.64 mmol) and TMSCl (0.4 ml, 3.17 mmol); (Z)-6a (70 mg, 0.19 mmol) with Ph₂CuLi (0.77 mmol) and TMSCl (0.3 ml, 2.37 mmol); and (E)-6b (80 mg, 0.19 mmol) with Me₂CuLi (0.37 mmol) and TMSCl (0.1 ml, 0.88 mmol); to give inseparable mixtures of (SiR*,R*,E)-9a, (SiR*,S*,E)-9a, (SiR*,R*,Z)-9a, and (SiR*,S*,Z)-9a of the compositions and in the yields given in Table 1. (Data of (SiR*,R*,E)-9a, (SiR*,S*,E)-9a not given, since these compounds arose only as components in the mixture of all four compounds of the type 9a.)

Data of (SiR^*,R^*,Z) -9a (from (Z)-6a, slightly contaminated with (SiR^*,S^*,Z) -9a): IR: 3080w, 3060w, 3020w, 2960s, 2930s, 2890s, 2850s, 1605m, 1485s, 1460m, 1455m, 1440m, 1410w, 1390w, 1375w, 1360w, 1265s, 1250s, 1200w, 1195w, 1115s, 1080s, 1070s, 1050w, 1035w, 1025m, 1020m, 1000w, 975w, 960m, 935m, 920w, 905m, 870s, 840s, 825s, 785m, 775m, 760s, 745s, 735s, 700s. ¹H-NMR: 7.75–7.08 (m, 15 arom. H); 4.58, 4.52 (AB, J_{AB} =12.2, PhC H_2 O); 3.58 (q, J=7.2, Me(Ph)CH); 3.56, 3.50 (AB, J_{AB} =12.5, SiCH₂O); 1.41 (d, J=7.2, Me(Ph)CH); 1.08 (s, t-Bu); -0.05 (s, MeSi); -0.38 (s, Me₃Si). ¹H, ¹H-NOE: irrad. at -0.38, responsive signals at 7.75–7.08, 4.58, 4.52, 3.56, 3.50, 1.08, and -0.05. ¹³C-NMR: 158.5 (s, PhC=); 145.2, 138.2, 137.7 (3s, 3 arom. C); 128.1, 127.4, 126.6, 126.5, 126.3, 126.1, 125.9, 125.7, 123.7 (9d, 15 arom. C); 115.5 (s, SiC=); 75.6 (t, PhCH₂O); 61.9 (t, SiCH₂O); 37.7 (d, Me(Ph)CH); 27.6 (q, Me_3 C); 18.3 (q, Me(Ph)CH); 17.4 (s, Me₃C); 0.0 (q, Me₃Si); -7.2 (q, MeSi). CI–MS: 517 [M+H]⁺.

3.4. (SiR*,R*,Z)- and (SiR*,S*,Z)-2-{[(Benzyloxy)methyl](tert-butyl)methylsilyl}-3-methyl-1-(tri-methylsiloxy)-1-phenylpent-1-ene (SiR*,R*,Z)-9b and (SiR*,S*,Z)-9b

According to 3.1, (E)-6a (100 mg, 0.37 mmol) was reacted with Et₂CuLi (1.64 mmol) and TMSCl (0.9 ml, 7.04 mmol); and (Z)-6a (100 mg, 0.27 mmol) with Et₂CuLi (1.10 mmol) and TMSCl (0.6 ml, 4.80 mmol); to give inseparable mixtures of (Si R^* , R^* ,Z)-9b and (Si R^* , S^* ,Z)-9b in the yields given in Table 1.

Data of (Si R^* , R^* ,Z)-9b (from (Z)-6a, slightly contaminated with (Si R^* , S^* ,Z)-9b): IR: 3060w, 3030w, 2960s, 2920s, 2890s, 2950s, 2810w, 1605m, 1575s, 1485m, 1460m, 1440m, 1410w, 1390w, 1375m, 1360w, 1310w, 1265s, 1250s, 1200w, 1170w, 1140w, 1100s, 1035w, 1025m, 1010w, 1000w, 980m, 965w, 950w, 920w, 905w, 875s, 840s, 825s, 785s, 760m, 735s, 700s. H-NMR: 7.37–7.15 (m, 10 arom. H); 4.56, 4.50 (AB, J_{AB} =12.2, PhC H_2 O); 3.56, 3.43 (AB, J_{AB} =12.7, SiCH₂O); 1.99–1.96 (m,

Et(Me)CH); 1.38-1.21, 1.09-1.06 (2m, MeCH₂); 1.05 (s, t-Bu); 0.89 (d, J=7.1, Et(Me)CH); 0.72 (t, J=7.4, MeCH₂); 0.32 (s, MeSi); -0.18 (s, Me₃Si). ¹H, ¹H-NOE: irrad. at -0.18, responsive signals at 7.37-7.15, 4.56, 4.50, 3.56, 3.43, 1.05, and 0.32; irrad. at 1.05, responsive signals at 7.37-7.15, 4.56, 4.50, 3.56, 3.43, 0.32, and -0.18. ¹³C-NMR: 157.2 (s, PhC=); 138.5, 138.3 (2s, 2 arom. C); 128.3 (d, arom. C); 126.8, 126.5 (2d, 2×2 arom. C); 126.4 (d, arom. C); 126.1, 125.7 (2d, 2×2 arom. C); 116.4 (s, SiC=); 75.6 (t, PhCH₂O); 62.3 (t, SiCH₂O); 36.5 (d, Et(Me)CH); 28.7 (t, MeCH₂); 27.4 (q, Me_3 C); 19.9 (q, Et(Me)CH); 17.7 (s, Me₃C); 11.9 (q, MeCH₂); 0.0 (q, Me₃Si); -6.2 (q, MeSi). CI–MS: 469 [M+H]⁺.

Data of (SiR^*,S^*,Z) -**9b** (from (Z)-**6a**, slightly contaminated with (SiR^*,R^*,Z) -**9b**): IR: 3060w, 3020m, 2950s, 2920s, 2890s, 2850s, 2810m, 1610m, 1570s, 1485m, 1460m, 1440m, 1410w, 1390w, 1375m, 1365m, 1310w, 1265s, 1250s, 1200w, 1170w, 1140w, 1100s, 1070s, 1025m, 1010w, 1000w, 985m, 950w, 920w, 905w, 875s, 840s, 825s, 780s, 760m, 735s, 700s. H-NMR: 7.36–7.14 (m, 10 arom. H); 4.54, 4.48 (AB, J_{AB} =12.2, PhC H_2 O); 3.51, 3.40 (AB, J_{AB} =12.7, SiCH₂O); 1.98–1.94 (m, Et(Me)CH); 1.35–1.13 (m, MeC H_2); 1.04 (s, t-Bu); 0.87 (d, J=7.1, Et(Me)CH); 0.71 (t, J=7.4, MeCH₂); 0.31 (s, MeSi); -0.19 (s, Me₃Si). H, H-NOE: irrad. at -0.19, responsive signals at 7.36–7.14, 4.54, 4.48, 3.51, 3.40, 1.04, and 0.31. H-NOE: 157.3 (s, PhC=); 138.5, 138.2 (2s, 2 arom. C); 128.4 (d, arom. C); 126.8, 126.5 (2d, 2×2 arom. C); 126.4 (d, arom. C); 126.1, 125.7 (2d, 2×2 arom. C); 116.7 (s, SiC=); 75.6 (t, PhCH₂O); 62.4 (t, SiCH₂O); 36.5 (d, Et(Me)CH); 28.5 (t, MeCH₂); 27.3 (q, Me_3 C); 19.9 (q, Et(Me)CH); 17.6 (s, Me₃C); 11.9 (q, MeCH₂); 0.0 (q, Me₃Si); -6.3 (q, MeSi). CI–MS: 469 [M+H]⁺.

3.5. (SiR*,R*,Z)- and (SiR*,S*,Z)-2-{[(Benzyloxy)methyl](tert-butyl)methylsilyl}-3-methyl-1-(tri-methylsiloxy)-1-phenylhept-1-ene (SiR*,R*,Z)-9c and (SiR*,S*,Z)-9c

According to 3.1, (E)-6a (100 mg, 0.27 mmol) was reacted with Bu₂CuLi (0.81 mmol) and TMSCl (0.25 ml, 1.97 mmol); and (Z)-6a (104 mg, 0.28 mmol) with Bu₂CuLi (0.57 mmol) and TMSCl (0.2 ml, 1.34 mmol); to give inseparable mixtures of (Si R^* , R^* ,Z)-9c and (Si R^* , S^* ,Z)-9c in the yields given in Table 1.

Data of (SiR*,S*,Z)-9c: IR: 3060w, 3020w, 2950s, 2920s, 2890s, 2850s, 1605w, 1580s, 1485w, 1460m, 1440m, 1410w, 1385w, 1375m, 1360w, 1310w, 1265s, 1250s, 1135w, 1100s, 1070s, 1025m, 1010w, 1000w, 980w, 920w, 905m, 890m, 860s, 840s, 825s, 780s, 760m, 730s, 600s. H-NMR: 7.32-7.10 (m, 10 arom. H); 4.50, 4.45 (AB, J_{AB} =6.1, PhC H_2 O); 3.48, 3.36 (AB, J_{AB} =12.7, SiC H_2 O); 2.03-1.99 (m, Bu(Me)CH); 1.27-1.02 (m, Me(C H_2)₃); 1.01 (s, t-Bu); 0.84 (d, J_2 =7.1, Bu(Me)CH); 0.76 (t, J_2 =6.9, MeC H_2); 0.27 (s, MeSi); -0.15 (s, Me₃Si). H, H-NOE: irrad. at -0.15, responsive signals at 7.32-7.10, 4.50, 4.45, 3.48, 3.36, 1.01, and 0.27. T3C-NMR: 157.1 (s, PhC=); 138.5, 138.2 (2s, 2 arom. C); 128.3 (d, arom. C); 126.8, 126.5 (2d, 2×2 arom. C); 126.4 (d, arom. C); 126.1, 125.7 (2d, 2×2 arom. C); 116.9 (s, SiC=); 75.6 (t, PhCH₂O); 62.4 (t, SiCH₂O); 35.5 (t, PrCH₂); 34.6 (d, Bu(Me)CH); 29.5 (t, EtCH₂); 27.3 (q, Me₃C); 21.4 (t, MeCH₂); 20.3 (d, Bu(Me)CH); 17.6 (s, (Me)₃C); 12.8 (q, MeCH₂); 0.0 (q, Me₃Si); -6.2 (q, MeSi). CI-MS: 497 [M+H]⁺.

3.6. (SiR*,R*,E)-, (SiR*,S*,E)-, (SiR*,R*,Z)-, and (SiR*,S*,Z)-2-{{(Benzyloxy)methyl}}(tert-butyl)-methylsilyl}-1-(trimethylsiloxy)-1,3-diphenylpent-1-ene(SiR*,R*,E)-9d, (SiR*,S*,E)-9d, (SiR*,R*,Z)-9d, and (SiR*,S*,Z)-9d

According to 3.1, (*E*)-**6b** (80 mg, 0.19 mmol) was reacted with Et₂CuLi (0.62 mmol) and TMSCl (0.1 ml, 0.88 mmol) to give (Si R^* , R^* ,E)-**9d**, (Si R^* , S^* ,E)-**9d**, (Si R^* , R^* ,Z)-**9d**, and (Si R^* , S^* ,Z)-**9d** (91 mg, 0.17 mmol, 89%) as an inseparable mixture. ¹H-NMR: rather complex; indicates the presence of the title structures. Structural proof is given with the derived products of the type **11d**.

3.7. (SiR*,R*,E)-, (SiR*,S*,E)-, (SiR*,R*,Z)-, and (SiR*,S*,Z)-2-{[(Benzyloxy)methyl](tert-butyl)-methylsilyl}-1-(trimethylsiloxy)-1,3-diphenylhept-1-ene SiR*,R*,E)-9e, (SiR*,S*,E)-9e, (SiR*,R*,Z)-9e, and (SiR*,S*,Z)-9e

According to 3.1, (*E*)-6b (80 mg, 0.19 mmol) was reacted with Bu₂CuLi (0.37 mmol) and TMSCl (0.1 ml, 0.88 mmol) to give (Si R^* , R^* ,E)-9e, (Si R^* , S^* ,E)-9e, (Si R^* , R^* ,Z)-9e, and (Si R^* , S^* ,Z)-9e (63 mg, 0.11 mmol, 59%) as an inseparable mixture. ¹H-NMR: rather complex; indicates the presence of the title structures. Structural proof is given with the derived products of the type 11e.

4. Hydrolysis of the enol ethers of the type 8 and 9

4.1. General procedure

To a soln. of the respective silyl enol ether in acetone (approx. 0.1 M) was added conc. H₂SO₄ (2–3 drops to 1.5 ml). The mixture was diluted with H₂O, neutralized with aq. NaHCO₃ soln., extracted with Et₂O, and chromatographed (SiO₂, hexane/EtOAc 25:1).

4.2. (1-[(tert-Butyl)dimethylsilyl]-2-phenylpropyl phenyl ketones 10 and 10'

According to 4.1, compounds of the type 8 gave equal amounts of the epimeric compounds 10 (first eluting) and 10' (second eluting) as colorless oils in 66-71% yield. The relative configurations of the stereogenic centers could not be elaborated.

Data of **10** (colorless oil): IR: 3080w, 3060w, 3020m, 2950s, 2920s, 2890m, 2880m, 2850s, 1655s, 1595m, 1575m, 1490m, 1460m, 1445m, 1410w, 1390w, 1360w, 1350m, 1330m, 1305w, 1285s, 1275s, 1250m, 1215m, 1205m, 1180m, 1155w, 1140w, 1070m, 1025w, 1015w, 1000s, 990s, 930w, 905w, 895w, 835s, 820s, 805s, 770m, 760s, 745s, 720m, 700s, 690s. ¹H-NMR: 7.82–7.15 (m, 10 arom. H); 3.78 (d, J=8.2, SiCH); 3.56–3.46 (m, Me(Ph)CH); 1.36 (d, J=6.8, Me(Ph)CH); 0.67 (s, t-Bu) –0.07, –0.17 (2s, Me₂Si). ¹³C-NMR: 203.8 (s, C=O); 147.2, 130.2, (2s, 2 arom. C); 132.2, 128.4, 128.3, 127.4, 126.3 (5d, 10 arom. C); 45.2 (d, SiCH); 41.3 (d, Me(Ph)CH); 27.0 (q, Me₃C); 22.4 (q, Me(Ph)CH); 17.9 (s, Me₃C); –5.2, –5.4 (2q, Me₂Si). CI–MS: 339 [M+H]⁺.

Data of **10**' (colorless crystals): M.p. 116–118°C (from oil). IR (CHCl₃): 3080w, 3060w, 3020w, 2969s, 2930s, 2890m, 2880m, 2850s, 1660s, 1595m, 1580m, 1490w, 1460m, 1450m, 1445m, 1410w, 1390w, 1375w, 1360m, 1330s, 1280s, 1260s, 1190m, 1180m, 1120m, 1090s, 1015s, 1000s, 985m, 935w, 900m, 830s, 700s. ¹H-NMR: 7.69–6.97 (m, 10 arom. H); 3.80 (d, J=11.2, SiCH); 3.63–3.53 (m, Me(Ph)CH); 1.43 (d, J=7.0, Me(Ph)CH); 0.94 (s, t-Bu) 0.20, -0.04 (2s, Me₂Si). ¹³C-NMR: 203.3 (s, C=O); 147.7, 139.4, (2s, 2 arom. C); 132.0, 128.1, 128.0, 126.9, 125.7 (5d, 10 arom. C); 46.8 (d, SiCH); 40.9 (d, Me(Ph)CH); 27.5 (q, Me₃C); 24.0 (q, Me(Ph)CH); 18.2 (s, Me₃C); -3.7, -5.0 (2q, Me₂Si). CI–MS: 339 [M+H]⁺.

4.3. (SiR*,3R*)- and (SiR*,3S*)-1-{[(Benzyloxy)methyl](tert-butyl)methylsilyl}-2-phenylpropyl phenyl ketones (SiR*,3R*)-11a, (SiR*,3S*)-11a, (SiR*,3R*)-11a'

According to 4.1, compounds of the type 9a gave inseparable mixtures of epimeric compounds $(SiR^*,3R^*)-11a/(SiR^*,3S^*)-11a$ (first eluting) and of $(SiR^*,3S^*)-11a'/(SiR^*,3S^*)-11a'$ (second eluting) as colorless oils. Yields, diastereomeric ratios, and diastereomeric excesses (in respect to the stereogenic centers at silicon and C(3)), see Table 1. Likewise, (-)-(SiR,3S)-11a (296 mg, 0.67 mmol,

84%) and (+)-(SiR,3S)-11a' (38 mg, 0.09 mmol, 11%) were obtained from (+)-(SiR,S,Z)-6a (290 mg, 0.79 mmol) by reaction with Ph₂CuLi (6 eq.) according to 3.1 and direct hydrolysis according to 4.1.

Data of (Si R^* ,3 R^*)-11a (from (Z)-6a, slightly contaminated with (Si R^* ,3 S^*)-11a): IR: 3080w, 3060w, 3020m, 2960s, 2920s, 2880m, 2750s, 2710w, 1655s, 1595m, 1525m, 1490m, 1460m, 1445s, 1390w, 1380m, 1360m, 1350m, 1325m, 1010w, 1285m, 1265s, 1250s, 1220s, 1205s, 1180m, 1155w, 1090s, 1070s, 1025m, 1020s, 1000m, 990m, 935w, 905w, 820s, 800m, 660s, 645s, 635s, 700s. H-NMR: 7.87–7.16 (m, 15 arom. H); 4.11, 4.05 (AB, J_{AB} =11.9, PhC H_2 O); 3.99 (d, J=7.9, SiCH); 3.63–3.50 (m, Me(Ph)CH); 3.14, 2.95 (AB, J_{AB} =12.9, SiCH₂O); 1.37 (d, J=6.9, Me(Ph)CH); 0.82 (s, t-Bu); 0.13 (s, MeSi). ¹³C-NMR: 203.6 (s, C=O); 147.0, 140.0, 138.8 (3s, 3 arom. C); 132.1, 128.4, 128.3, 128.1, 127.9, 127.3, 127.0, 126.3 (8d, 15 arom. C); 76.7 (t, PhCH₂O); 61.0 (t, SiCH₂O); 43.6 (d, SiCH); 40.9 (d, Me(Ph)CH); 27.5 (q, Me_3 C); 22.2 (q, Me(Ph)CH); 18.1 (s, Me₃C); -8.3 (q, MeSi). CI–MS: 445 [M+H]⁺.

Data of $(SiR^*,3R^*)$ -11a' (from (Z)-6a, slightly contaminated with $(SiR^*,3S^*)$ -11a'): IR: 3080w, 3060m, 3020m, 2960s, 2930s, 2880s, 2850s, 2810w, 1660s, 1595m, 1575m, 1490m, 1465m, 1455s, 1445s, 1404w, 1390w, 1380m, 1360m, 1325m, 1280m, 1260s, 1210m, 1190m, 1180m, 1155m, 1090s, 1070s, 1025m, 1015m, 1000m, 985w, 935w, 905m, 830s, 805m, 785m, 760s, 700s. ¹H-NMR: 7.94–6.94 (m, 15 arom. H); 4.31 (s, PhCH₂O); 3.89 (d, J=11.1, SiCH); 3.71–3.60 (m, Me(Ph)CH); 3.32, 3.22 (AB, J_{AB} =12.9, SiCH₂O); 1.45 (d, J=7.0, Me(Ph)CH); 0.97 (s, t-Bu); 0.12 (s, MeSi). ¹³C-NMR: 203.3 (s, C=O); 147.5, 139.2, 138.6 (3s, 3 arom. C); 131.9, 128.2, 128.1, 128.0, 127.5, 127.3, 127.0, 125.8 (8d, 15 arom. C); 77.0 (t, PhCH₂O); 61.0 (t, SiCH₂O); 46.3 (d, SiCH); 41.0 (d, Me(Ph)CH); 27.8 (q, Me_3 C); 23.7 (q, Me(Ph)CH); 18.7 (s, Me₃C); -6.4 (q, MeSi). CI–MS: 445 [M+H]⁺.

Data of (-)-(SiR,3S)-11a (from (R,E)-6a, slightly contaminated with (SiR,3R)-11a): [α]_D²³=-13.8 (c=1.0, THF). IR: 3080w, 3060m, 3020m, 2960s, 2930s, 2880s, 2850s, 2810m, 1660s, 1595m, 1575m, 1490m, 1460m, 1450s, 1445s, 1430w, 1390w, 1380m, 1360m, 1350m, 1330m, 1310w, 1285m, 1265s, 1250s, 1205s, 1180m, 1155w, 1090m, 1070s, 1025m, 1000s, 995s, 935w, 905m, 825s, 805s, 780m, 760s, 750s, 735s, 700s. ¹H-NMR: 7.87-7.11 (m, 15 arom. H); 4.24, 4.18 (AB, J_{AB} =11.9, PhC H_2 O); 4.01 (d, J=8.5, SiCH); 3.66-3.56 (m, Me(Ph)CH); 2.97, 2.64 (AB, J_{AB} =12.8, SiCH₂O); 1.33 (d, J=6.9, Me(Ph)CH); 0.76 (s, t-Bu) 0.13 (s, MeSi). ¹³C-NMR: 203.8 (s, C=O); 146.9, 139.9, 138.8 (3s, 3 arom. C); 132.3, 128.5, 128.3, 128.1, 127.4, 127.2, 126.3 (7d, 15 arom. C); 76.9 (t, PhCH₂O); 60.7 (t, SiCH₂O); 43.2 (d, SiCH); 41.1 (d, Me(Ph)CH); 27.5 (q, Me_3 C); 22.5 (q, Me(Ph)CH); 18.2 (s, Me₃C); -8.3 (q, MeSi). CI-MS: 445 [M+H]⁺.

Data of (+)-(Si*R*,3*S*)-11a' (from (*R*,*E*)-6a, slightly contaminated with (Si*R*,3*R*)-11a'): $[\alpha]_D^{23}$ =33.6 (c=1.2, THF). IR: 3080w, 3060w, 3020m, 2960s, 2920s, 2850s, 2810w, 1725w, 1685w, 1660s, 1595m, 1580m, 1490m, 1470m, 1450s, 1445s, 1390w, 1375m, 1360m, 1325m, 1280m, 1260s, 1210m, 1190m, 1180m, 1155w, 1090s, 1070s, 1025m, 1015m, 1000m, 985m, 935w, 900w, 830s, 800m, 790m, 760s, 735s, 700s. ¹H-NMR: 7.69–6.91 (*m*, 15 arom. H); 4.29, 4.23 (*AB*, J_{AB} =11.8, PhC H_2 O); 3.94 (*d*, J_2 =11.2, SiCH) 3.66–3.55 (*m*, Me(Ph)CH); 3.30, 3.13 (*AB*, J_{AB} =12.8, SiCH₂O); 1.44 (*d*, J_2 =6.8, *Me*(Ph)CH); 1.04 (*s*, *t*-Bu); 0.11 (*s*, MeSi). ¹³C-NMR: 203.0 (*s*, C=O); 147.3, 139.2, 138.6 (3*s*, 3 arom. C); 131.8, 128.0, 127.9, 127.8, 127.4, 127.1, 127.0, 126.0 (8*d*, 15 arom. C); 76.9 (*t*, PhCH₂O); 61.7 (*t*, SiCH₂O); 45.6 (*d*, SiCH); 41.1 (*d*, Me(Ph)CH); 28.0 (*q*, *Me*₃C); 23.7 (*q*, *Me*(Ph)CH); 18.6 (*s*, Me₃C); -7.5 (*q*, MeSi). CI–MS: 445 [*M*+H]⁺.

4.4. (SiR*,3R*)- and (SiR*,3S*)-1- $\{[(Benzyloxy)methyl]$ (tert-butyl)methylsilyl}-2-methylbutyl phenyl ketones (SiR*,3R*)-11b, $(Si^R*,3^S*)$ -11b, $(Si^R*,3^R*)$ -11b', and (SiR*,3S*)-11b'

According to 4.1, compounds of the type **9b** gave inseparable mixtures of $(SiR^*,3R^*)-11b$, $(SiR^*,3S^*)-11b$, $(SiR^*,3R^*)-11b'$, and $(SiR^*,3S^*)-11b'$ as colorless oils. Yields, diastereomeric ratios, and diastereomeric excesses (in respect to the stereogenic centers at silicon and C(3)), see Table 1.

Data of mixture $(SiR^*,3R^*)-11b/(SiR^*,3R^*)-11b'$ (from (Z)-6a, slightly contaminated with $(SiR^*,3S^*)-11b/(SiR^*,3S^*)-11b'$): IR: 3080w, 3060m, 3030m, 2960s, 2930s, 2850s, 2810m, 1730w,

1660s, 1595m, 1575m, 1490w, 1460s, 1445s, 1430m, 1410w, 1390m, 1380m, 1360m, 1320m, 1300w, 1280m, 1255s, 1210s, 1205s, 1180m, 1155m, 1095s, 1070s, 1035m, 1000m, 990m, 935w, 905m, 825s, 805s, 780m, 735s, 700s. ¹H-NMR (major two isomers): 7.87–7.83, 7.43–7.04 (2m, 10 arom. H); 4.23, 4.05, 4.01 (s and AB, J_{AB} =12.2, PhC H_2 O); 3.62, 3.44 (2d, J=5.6, 7.9, SiCH); 3.19, 3.11, 3.14, 2.89 (2AB, J_{AB} =12.9, 12.9, SiCH $_2$ O); 2.23–2.10, 2.07–1.98 (2m, Et(Me)CH); 1.55–1.39, 1.20–1.05 (2m, MeC H_2); 1.00, 0.99 (2d, J=6.8, 6.8, Et(Me)CH); 0.91, 0.81 (2s, t-Bu); 0.81, 0.72 (2t, J=7.8, 7.4, MeCH $_2$); 0.03, 0.00 (2s, MeSi). ¹³C-NMR (major two isomers): 204.0, 203.5 (2s, C=O); 140.1, 139.9, 138.8, 138.7 (4s, 2 arom. C); 132.2, 132.1, 128.4, 128.3, 128.2, 128.1, 128.0, 127.5, 127.2, 127.0 (10d, 10 arom. C); 77.9, 76.8 (2t, PhC H_2 O); 61.6, 61.1 (2t, SiCH $_2$ O); 44.0, 41.3 (2d, SiCH); 36.4, 36.2 (2d, Et(Me)CH); 31.1, 29.2 (2t, MeC H_2); 27.7 (q, Me_3 C); 20.1, 18.6 (2q, MeCH); 18.4, 18.2 (2s, Me $_3$ C); 12.2, 11.7 (2q, MeCH $_2$); -6.8, -8.3 (2q, MeSi). ²⁹Si-NMR: 4.92, 4.26, 4.25, 3.87 (ratio 3:38:58:1). CI–MS: 397 [M+H] $^+$.

4.5. (SiR*,3R*)- and (SiR*,3S*)-1-{[(Benzyloxy)methyl](tert-butyl)methylsilyl}-2-methylhexyl phenyl ketones (SiR*,3R*)-11c, (SiR*,3S*)-11c, (SiR*,3R*)-11c', and (SiR*,3S*)-11c'

According to 4.1, compounds of the type 9c gave inseparable mixtures of $(SiR^*,3R^*)-11c$, $(SiR^*,3S^*)-11c'$, and $(SiR^*,3S^*)-11c'$ as colorless oils. Yields, diastereomeric ratios, and diastereomeric excesses (in respect to the stereogenic centers at silicon and C(3)), see Table 1.

Data of mixture $(SiR^*,3S^*)$ -11c/ $(SiR^*,3S^*)$ -11c' (from (*E*)-6a, slightly contaminated with $(SiR^*,3R^*)$ -11c/ $(SiR^*,3R^*)$ -11c'): IR: 3080w, 3060m, 3020m, 2850s, 2820s, 2790s, 2810m, 1660s, 1595m, 1575m, 1490w, 1460s, 1445s, 1375m, 1360m, 1320m, 1275m, 1250s, 1220m, 1210m, 1180m,

1155w, 1090m, 1070s, 1025m, 1000m, 995m, 935w, 905w, 825s, 800m, 780m, 695s. ¹H-NMR (major two compounds): 7.93–7.89, 7.49–7.11 (2m, 10 arom. H); 4.36, 4.16, 4.12 (s and AB, J=11.5, PhCH₂O); 3.59, 3.57 (2d, J=6.1, 7.8, SiCH); 3.31, 3.24, 3.23, 3.02 (2AB, J_{AB}=12.9, 12.9, SiCH₂O); 2.26–2.19, (m, Bu(Me)CH); 1.55–1.19 (m, Me(CH₂)₃); 1.05, 1.03 (2d, J=5.7, 6.7, MeCH); 0.97, 0.88 (2s, t-Bu); 0.83, 0.81 (2t, J=6.8, 7.6, MeCH₂); 0.08, 0.06 (2s, MeSi). ¹³C-NMR (major two compounds): 203.9, 203.7 (s, C=O); 140.1, 140.0, 138.8, 138.7 (4s, 2 arom. C); 132.2, 132.1, 128.3, 128.24, 128.2, 128.1, 128.0, 127.5, 127.3, 127.1 (10d, 10 arom. C); 77.1, 76.8 (2t, PhCH₂O); 61.7, 61.0 (2t, SiCH₂O); 43.6, 42.3 (2d, SiCH); 38.1, 35.9 (2t, PrCH₂); 34.7, 34.5 (2d, Bu(Me)CH); 29.9, 29.8 (2t, EtCH₂); 27.9, 27.7 (2q, Me₃C); 22.7 (t, MeCH₂); 20.8, 19.2 (2q, MeCH); 18.4, 18.3 (2s, Me₃C); 14.0 (q, MeCH₂); -7.3, -7.9 (2q, MeSi). ²⁹Si-NMR: 4.93, 4.28, 4.24, 3.89 (ratio 63:5:4:28). CI–MS: 425 [M+H]⁺.

4.6. (SiR*,3R*)- and (SiR*,3S*)-1-{[(Benzyloxy)methyl](tert-butyl)methylsilyl}-2-phenylbutyl phenyl ketones (SiR*,3R*)-11d, (SiR*,3S*)-11d, (SiR*,3R*)-11d', and (SiR*,3S*)-11d'

According to 4.1, compounds **9d** (91 mg, 0.17 mmol, 89%) gave inseparable mixtures of epimeric (Si R^* ,3 R^*)-**11d**/(Si R^* ,3 S^*)-**11d** (25 mg, 0.05 mmol, 32%, first eluting, ratio 5:1, determined by ¹H-NMR) and of (Si R^* ,3 R^*)-**11d**'/(Si R^* ,3 S^*)-**11d**' (32 mg, 0.07 mmol, 41%, second eluting, ratio 1.6:1, determined by ¹H-NMR).

Data of $(SiR^*,3R^*)$ -11d/ $(SiR^*,3S^*)$ -11d (as a mixture): IR: 3080w, 3060m, 3020m, 2980s, 2930s, 2850s, 2810m, 1660s, 1595m, 1575m, 1490m, 1460m, 1450s, 1445s, 1430w, 1390w, 1375m, 1360m, 1325m, 1310w, 1290w, 1260s, 1250s, 1215m, 1200m, 1180m, 1155w, 1140w, 1105m, 1090s, 1070s, 1025m, 1010m, 1000m, 985m, 960w, 935w, 905w, 860m, 830m, 800m, 780m, 755m, 740s, 700s. ¹H-NMR: 7.94–7.82, 7.56–7.12 (2m, 15 arom. H); 4.20, 4.14, 4.11, 4.05 (2AB, J_{AB} =11.9, 9.3, PhC H_2 O); 4.05, 4.02 (2d, J=9.3, 10.6, SiCH); 3.35–3.25 (m, Ph(Et)CH); 3.13, 2.97, 2.90, 2.47 (2AB, J_{AB} =12.9, 12.8, SiCH $_2$ O); 1.90–1.77, 1.70–1.57 (2m, MeC H_2); 0.79, 0.75 (2s, t-Bu) 0.59, 0.57 (2t, J=5.9, 7.2 MeCH $_2$); -0.02, -0.07 (2s, MeSi). ¹³C-NMR: 204.1, 204.0 (2s, C=O); 144.1, 139.9, 138.8 (3s, 3 arom. C); 132.4, 132.2, 128.6, 128.5, 128.3, 128.2, 128.1, 127.9, 127.4, 127.3, 127.2, 126.4 (12d, 15 arom. C); 76.8, 76.7 (2t, PhCH $_2$ O); 61.0, 60.1 (2t, SiCH $_2$ O); 48.7, 48.6 (2d, SiCH); 44.3, 43.1 (2d, Ph(Et)CH); 29.2, 28.7 (2t, MeCH $_2$); 27.6 (q, Me3C); 18.3, 18.1 (2s, Me $_3$ C); 12.3 (q, MeCH $_2$); -8.4 (q, MeSi). CI–MS: 459 [M+H] $^+$.

Data of (Si*R**,3*R**)-11d' and (Si*R**,3*S**)-11d' (as a mixture): IR: 3080*m*, 3060*s*, 3020*s*, 2950*s*, 2920*s*, 2850*s*, 2810*m*, 1655*s*, 1595*m*, 1575*m*, 1490*m*, 1460*s*, 1450*s*, 1430*m*, 1405*m*, 1390*m*, 1375*m*, 1360*m*, 1320*m*, 1300*m*, 1255*s*, 1215*m*, 1185*m*, 1155*m*, 1105*m*, 1090*s*, 1070*s*, 1025*m*, 1010*m*, 1000*m*, 975*m*, 930*m*, 920*m*, 900*m*, 845*m*, 835*m*, 800*m*, 780*m*, 765*m*, 735*m*, 700*s*. ¹H-NMR: 7.68–7.62, 7.42–6.95 (2*m*, 15 arom. H); 4.33, 4.31, 4.25 (*s* and *AB*, *J_{AB}*=12.0, PhC*H*₂O); 4.00, 3.93 (2*d*, *J*=11.1, 11.0, SiCH); 3.39–3.24 (*m*, Ph(Et)C*H*); 3.34, 3.24, 3.31, 3.14 (2*AB*, *J_{AB}*=12.9, 12.8, SiCH₂O); 2.15–1.99, 1.70–1.58 (2*m*, MeC*H*₂); 1.05, 0.98 (2*s*, *t*-Bu) 0.75, 0.66 (2*t*, *J*=12.3, 7.4 *Me*CH₂); 0.10 (*s*, MeSi). ¹³C-NMR: 203.1, 202.8 (2*s*, C=O); 144.6, 144.4, 139.4, 138.6 (4*s*, 3 arom. C); 131.8, 128.6, 128.5, 128.3, 128.2, 128.1, 128.9, 127.9, 127.7, 127.5, 127.4, 127.3, 127.1, 125.7 (14*d*, 15 arom. C); 77.1, 76.9 (2*t*, PhCH₂O); 61.8, 61.0 (2*t*, SiCH₂O); 48.4, 48.2 (2*d*, SiCH); 45.9, 45.0 (2*d*, Ph(Et)*C*H); 29.7, 29.1 (2*t*, MeCH₂); 28.1, 27.8 (*q*, *Me*₃C); 18.7, 18.1 (2*s*, Me₃C); 12.3, 12.2 (2*q*, *Me*CH₂); -6.3, -7.6 (*q*, MeSi). CI–MS: 459 [*M*+H]*.

4.7. (SiR*,3R*)- and (SiR*,3S*)-1-{[(Benzyloxy)methyl](tert-butyl)methylsilyl}-2-phenylhexyl phenyl ketones (SiR*,3R*)-11e, (SiR*,3S*)-11e, (SiR*,3R*)-11e', and (SiR*,3S*)-11e'

According to 4.1, compounds **9e** (63 mg, 0.11 mmol, 59%) gave inseparable mixtures of epimeric (Si R^* ,3 R^*)-**11e**/(Si R^* ,3 S^*)-**11e** (17 mg, 0.03 mmol, 27%, first eluting, ratio 8:1, determined by ¹H-NMR) and of (Si R^* ,3 R^*)-**11e**' and (Si R^* ,3 S^*)-**11e**' (36 mg, 0.07 mmol, 66%, second eluting, ratio 3:1, determined by ¹H-NMR).

Data of $(SiR^*,3R^*)$ -11e/ $(SiR^*,3S^*)$ -11e (as a mixture): IR: 3080m, 3060m, 3020m, 2950s, 2930s, 2850s, 2810m, 1665s, 1595m, 1580m, 1490m, 1465s, 1450s, 1445s, 1410w, 1390w, 1375m, 1360m, 1325m, 1300w, 1275m, 1259s, 1205m, 1185m, 1155w, 1105m, 1090s, 1070s, 1025m, 1000m, 985m, 935w, 905m, 825s, 800m, 780m, 765s, 735s, 700s. H-NMR (major isomer): 7.80–7.77, 7.45–7.10 (2m, 15 arom. H); 4.10, 4.03 (AB, J_{AB} =11.9, PhC H_2 O); 3.97 (d, J=8.1, SiCH); 3.35–3.28 (m, Ph(Bu)CH); 3.09, 2.94 (AB, J_{AB} =12.9, SiCH₂O); 1.83–1.64 (m, PrC H_2); 1.21–0.80 (m, Me(C H_2)₂); 0.78 (s, t-Bu) 0.71 (t, J=7.4, MeCH₂); -0.08 (s, MeSi). ¹³C-NMR (major isomer): 203.8 (s, C=O); 144.8, 140.0, 138.8 (3s, 3 arom. C); 132.1, 128.4, 128.3, 128.2, 127.9, 127.3, 127.0, 126.4 (8d, 15 arom. C); 76.6 (t, PhC H_2 O); 61.0 (t, SiCH₂O); 46.7 (d, SiCH); 44.0 (d, Ph(Bu)CH); 35.3 (t, PrC H_2); 30.3 (t, EtCH₂); 27.5 (q, Me3C); 22.4 (t, MeC H_2); 15.2 (s, Me₃C); 13.9 (q, MeC H_2); -8.2 (q, MeSi). CI-MS: 487 [M+H]⁺.

Data of (Si*R**,3*R**)-**11e**′ and (Si*R**,3*S**)-**11e**′ (as a mixture): IR: 3080*w*, 3060*m*, 3020*m*, 2950*s*, 2930*s*, 2850*s*, 2810*m*, 1660*s*, 1595*m*, 1580*m*, 1490*m*, 1465*m*, 1450*m*, 1445*m*, 1390*w*, 1375*m*, 1360*m*, 1325*m*, 1300*w*, 1280*m*, 1250*m*, 1215*m*, 1200*m*, 1180*m*, 1165*w*, 1090*m*, 1070*m*, 1025*w*, 1000*m*, 940*w*, 905*w*, 825*m*, 800*m*, 780*m*, 735*m*, 700*s*. ¹H-NMR: 7.67–7.61, 7.41–6.94 (2*m*, 15 arom. H); 4.33, 4.26 (2*s*, PhC*H*₂O); 3.98, 3.92 (2*d*, *J*=11.3, 11.0, SiCH); 3.48–3.34 (*m*, Ph(Bu)C*H*); 3.34, 3.24, 3.24, 3.12 (2*AB*, *J_{AB}*=13.0, 12.8, SiCH₂O); 2.06–1.95, 1.69–1.56 (2*m*, PrC*H*₂); 1.31–1.10 (*m*, Me(C*H*₂)₂); 1.05, 0.98 (2*s*, *t*-Bu) 0.81, 0.76 (2*t*, *J*=7.4, 7.2, *Me*CH₂); 0.11, 0.10 (2*s*, MeSi). ¹³C-NMR: 203.2, 202.9 (2*s*, C=O); 145.0, 144.8, 139.4, 138.6 (4*s*, 3 arom. C); 131.8, 128.2, 128.0, 127.9, 127.7, 127.6, 127.4, 127.3, 127.1, 125.7 (10*d*, 15 arom. C); 77.1, 76.9 (2*t*, PhCH₂O); 61.9, 61.0 (2*t*, SiCH₂O); 46.7, 46.5 (2*d*, SiCH); 46.1, 45.2 (2*d*, Ph(Bu)CH); 36.6, 36.5 (2*t*, PrCH₂); 29.9, 29.8 (2*t*, EtCH₂); 28.0, 27.9 (2*q*, *Me*₃C); 22.6 (*t*, MeCH₂); 18.6 (*s*, Me₃C); 15.2, 14.0 (2*q*, *Me*CH₂); -6.2, -7.6 (2*q*, MeSi). CI–MS: 487 [*M*+H]⁺.

5. Removal of the silicon groups of compounds of the type 10 and 11

5.1. General procedure

To a soln, of the respective α -silylated ketone in CH₃CN (0.1 M) was added TBAF (4 eq., 1 M in THF). The mixture was stirred for 1 h and diluted with H₂O. It was extracted with Et₂O and chromatographed (SiO₂, hexane/EtOAc 25:1).

5.2. (\pm) - and (-)-(R)-Phenyl 2-phenylpropyl ketone (\pm) -12a and (-)-(R)-12a

According to 5.1, a mixture of all racemic isomeric compounds of the type **10** (159 mg, 0.47 mmol) gave (\pm)-**12a** (86 mg, 0.38 mmol, 82%) as a colorless oil. Likewise, a mixture of all racemic compounds of the type **11** (170 mg, 0.38 mmol) gave (\pm)-**12a** (75 mg, 0.33 mmol, 87%) (oil), and (-)-(SiR,3S)-**11a** (170 mg, 0.38 mmol) gave (-)-(R)-**12a** (75 mg, 0.33 mmol, 87%) as a colorless solid. M.p.: 45–50°C (oil for (-)-(R)-**12a**, ²² 45–48°C for (+)-(S)-**12a**²³). [α]_D²³=-12.4 (c=1.4, CCl₄)(-14.8 (c=1.2, CCl₄) for (-)-(R)-**12a**, ²² +13.3 (c=2.8, CCl₄) for (+)-(S)-**12a**²³). IR: 3080m, 3050m, 3020s, 2960s, 2920m, 2870m, 1685s, 1595s, 1580m, 1490m, 1445s, 1400m, 1360m, 1310m, 1270s, 1215s, 1200s, 1175m, 1155s, 1100s, 1075s, 1020s, 990s, 935s, 905s, 875s, 840s, 835s, 790s, 755s, 700s. ¹H-NMR: 7.98–6.96 (s, 10 arom. H); 3.55–3.42 (s, Me(Ph)CH); 3.28, 3.16 (s) of ABX, s, s) 13.2, 126.5, 128.0, 126.8, 126.2 (5s), 10 arom. C); 47.0 (s, PhCOCH₂); 35.5 (s, MeCH); 15.2 (s, MeCH). CI–MS: 242 [s+NH₄]+.

The enantiomeric excess of (-)-(R)-12a was determined by means of the Mosher method:²⁴ (-)-(R)-12a (160 mg, 0.71 mmol) was reduced with LiAlH₄ (32 mg, 0.84 mmol) in Et₂O (5 ml) to give two isomeric alcohols (ratio ca. 1:1, 94%), which were separated by chromatography (SiO₂, hexane/EtOAc 25:1). The first eluting isomer (20 mg, 0.09 mmol) was transferred into the respective Mosher ester (20 mg, 0.05 mmol, 61%) by treatment with (-)-(R)- α -methoxy- α -trifluoromethylphenylacetic acid chloride (112 mg, 0.44 mmol) in the presence of Et₃N. ¹⁹F-NMR: 71.42, 71.91 (ratio 97:3).

5.3. 2-Methylbutyl phenyl ketone 12b

According to 5.1, a mixture of all compounds of the type **11b** (243 mg, 0.61 mmol) gave **12b** (93 mg, 0.53 mmol, 86%) as a colorless oil. IR: 3080w, 3060m, 3020w, 2960s, 2920s, 2870s, 1690s, 1595m, 1580m, 1470m, 1445s, 1405m, 1360s, 1320m, 1280m, 1260m, 1205s, 1180m, 1155w, 1100w, 1075w, 1015m, 1000m, 995w, 965w, 930w, 915w, 890w, 840w, 825w, 780w, 750s, 690s. ¹H-NMR: 7.78-7.24 (m, 5 arom. H); 2.76, 2.55 (AB of ABX, $J_{AB}=15.7$, $J_{AX}=7.9$, $J_{BX}=5.7$, PhCOC H_2); 1.96-1.85 (m, Me(Et)CH); 1.32-0.98 (m, MeC H_2); 0.77 (d, J=6.7, MeCH); 0.74 (t, J=7.4, MeCH₂). ¹³C-NMR: 200.4 (s, C=O); 137.4 (s, arom. C); 132.7 (d, arom. C); 128.4, 128.0 (2d, 2×2 arom. C); 45.5 (t, PhCOC H_2); 31.3 (d, Me(Et)CH); 29.6 (t, MeC H_2): 19.4 (q, MeCH); 11.3 (q, MeCH₂). CI-MS: 194 ($M+NH_4$)⁺.

5.4. 2-Methylhexyl phenyl ketone 12c

According to 5.1, a mixture of all compounds of the type **11c** (261 mg, 0.62 mmol) gave **12c** (112 mg, 0.55 mmol, 89%) as a colorless oil. IR: 3080w, 3060w, 3020w, 2950s, 2920s, 2870s, 2850s, 1690s, 1595m, 1580m, 1460m, 1445s, 1405w, 1375m, 1360m, 1315w, 1280m, 1250w, 1215m, 1190w, 1180m, 1155w, 1100w, 1070w, 1000m, 940w, 910w, 890w, 840w, 770w, 750s, 690s. ¹H-NMR: 7.78–7.23 (m, 5 arom. H); 2.76, 2.55 (AB of ABX, J_{AB} =15.8, J_{AX} =7.9, J_{BX} =5.7, PhCOCH₂); 2.00–1.93 (m, Me(Bu)CH); 1.24–0.99 (m, Me(CH₂)₃); 0.77 (d, J=6.7, MeCH); 0.76–0.68 (m, MeCH₂). ¹³C-NMR: 200.4 (s, C=O); 137.4 (s, arom. C); 132.7 (d, arom. C); 128.4, 128.0 (2d, 2×2 arom. C); 45.9 (t, PhCOCH₂); 36.8 (t, PrCH₂); 29.7 (d, Me(Bu)CH); 29.2 (t, EtCH₂); 22.8 (t, MeCH₂); 19.9 (t, t), t0.40 (t0, t1.40 (t0, t1.51 (t1.51 (t2.52) (t1.51 (t3.53 (t5.54 (t5.54 (t5.55

5.6. Phenyl 2-phenylbutyl ketone 12d

According to 5.1, a mixture of all compounds of the type 11d (50 mg, 0.11 mmol) gave 12d (20 mg, 0.08 mmol, 77%) as a colorless oil. IR: 3080m, 3060m, 3020m, 2960s, 2920s, 2870m, 1725w, 1690s, 1595m, 1580m, 1490m, 1445s, 1405m, 1375m, 1365m, 1350m, 1330m, 1315m, 1280m, 1245m, 1210m, 1200m, 1180m, 1155w, 1100w, 1070w, 1115m, 975m, 950w, 925w, 905w, 840w, 785w, 750s, 700s. 1 H-NMR: 7.94-7.16 (m, 10 arom. H); 3.30-3.23 (2m, Ph(Et)CH and PhCOCH₂); 1.88-1.59 (m, MeCH₂); 0.82 (t, J=7.3, MeCH₂). 13 C-NMR: 199.1 (s, C=O); 144.6, 137.2 (2s, 2 arom. C); 132.8, 128.4, 128.3, 128.0, 127.6 (5d, 5×2 arom. C); 45.5 (t, PhCOCH₂); 43.0 (d, Ph(Et)CH); 29.1 (t, MeCH₂); 12.0 (g, MeCH₂). CI-MS: 256 [M+NH₄] $^{+}$.

5.7. Phenyl 2-phenylhexyl ketone 12e

According to 5.1, a mixture of all compounds of the type **11e** (50 mg, 0.10 mmol) gave **12e** (20 mg, 0.08 mmol, 73%) as a colorless oil. IR: 3080*m*, 3050*m*, 3020*s*, 3000*m*, 2950*s*, 2920*s*, 2850*s*, 1725*w*, 1690*s*, 1595*s*, 1580*m*, 1490*m*, 1465*m*, 1445*m*, 1405*m*, 1365*m*, 1350*m*, 1330*m*, 1315*m*, 1285*m*, 1270*m*, 1250*m*, 1215*m*, 1200*m*, 1180*m*, 1155*w*, 1110*w*, 1070*m*, 1020*m*, 975*m*, 940*w*, 930*w*, 905*w*, 885*w*, 870*w*, 840*w*, 750*s*, 725*m*, 700*s*. ¹H-NMR: 7.95–7.15 (*m*, 10 arom. H); 3.38–3.31 (*m*, Ph(Bu)CH); 3.29–3.21 (*m*, PhCOCH₂); 1.80–1.58 (*m*, PrCH₂); 1.35–1.07 (*m*, Me(CH₂)₂); 0.83 (*t*, *J*=7.2, *Me*CH₂). ¹³C-NMR: 198.9 (*s*, C=O); 144.7, 137.0 (2*s*, 2 arom. C); 132.5, 128.2, 128.1, 128.0, 127.2 (5*d*, 5×2 arom. C); 45.7 (*t*, PhCOCH₂); 41.0 (*d*, Ph(Bu)CH); 35.7 (*t*, PrCH₂); 29.3 (*t*, EtCH₂); 22.3 (*t*, MeCH₂); 13.6 (*q*, *Me*CH₂). CI–MS: 284 [*M*+NH₄]⁺.

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